Supplementary material to

The $v_{10} = 1$ level of Propyne, $H_3C-C=CH$, and Its Interactions with $v_9 = 1$ and $v_{10} = 2$

by H. S. P. Müller, P. Pracna, and V.-M. Horneman, in *J. Mol. Spectrosc.* **216** (2002) 397–407.

This file contains data obtained with Pickett's SPFIT program which was written to fit general asymmetric top molecules. Table 1 contains the v = 0 constants. Options exist to fit *l*doubled states of a symmetric rotor and to consider vibration-rotation interaction. The fit differs slightly from that presented in the paper which was obtained employing a program written by Dr. Petr Pracna to fit general symmetric top molecules. First, instead of fitting rotational and centrifugal distortion constants for all states, in the present case differences of these constants from those of the v = 0 state have been fit. This is a rather technical difference with no effect on the quality of the fit. However, there are also slight differences in the type and number of parameters determined as well as in the weighting scheme. Table 2 presents the $v_9 = 1$ and v_{10} = 2 constants used in the combined fit while **Table 3** gives the resulting constants for the v_{10} = 1 state. As can be seen in the second part of this file (Tables 5 to 7), the resulting differences are usually small but not always insignificant. In the $v_{10} = 1$ fit, the v = 0 constants were kept fixed, and the uncertainties of the ground state parameters have not been propagated. This is a usually done. Table 4 demonstrates that inclusion of the ground states lines in the fit and release of the ground state constants mostly changes both the constants and their uncertainties only slightly. The uncertainties of the $v_{10} = 1$ parameters A - B, $A\zeta$, and η_K increase by a larger amount. Only the value of L_{KKI} changes by more than 1σ .

IMPORTANT NOTE: The $v_{10} = 1$ files provided in the DATA or FITTING SPECTRA sections will work only with the 7 July 2001 and 26 Sept. 2001 versions of the program !! In previous versions there was a bug that allowed one *l*-doubled state to be fit only.

Table 1Spectroscopic constants (MHz) of propyne, $H_3C-C=CH$, in its ground vibrational state.

Parameter	Value
A-B	150 594.45 (46)
В	8 545.876 862 1 (217)
$D_J \cdot 10^3$	2.939 241 0 (251)
$D_{JK} \cdot 10^3$	163.41449 (209)
D_K	2.9072 (122)
$H_J \cdot 10^{12}$	-63.2 (61)
$H_{JK} \cdot 10^9$	914.08 (89)
$H_{\rm KJ}\cdot 10^6$	5.3021 (114)
$L_{JJK} \cdot 10^{12}$	-6.313 (148)
$L_{JK} \cdot 10^{12}$	-41.81 (189)
$L_{KKJ} \cdot 10^{12}$	-419.4 (203)

Spectroscopic constants^a (MHz) of propyne, $H_3C-C=CH$, in its $v_9 = 1$ and $v_{10} = 2$ vibrationally excited states used for the global $v_{10} = 1$ fit.

	$v_9 = 1$	$v_{10} = 2, l = 0$	$v_{10} = 2, \ l = \pm 2$
$E_{\rm v}\cdot 10^{-6}$	19.143 863 85	19.51235446	20.126 227 48
$\Delta(A-B)$	-45.181	-140.787	-206.169
ΔB	5.18179	47.82659	47.38337
$\Delta D_J \cdot 10^6$	12.39	161.8	153.08
$\Delta D_{JK} \cdot 10^6$	91.3	686.6	884.7
$\Delta D_{K} \cdot 10^{3}$	23.3	86.2	79.7
q_{22}	-9.01900	-16.78449^{b}	b
$q^{J}_{22} \cdot 10^{6}$	19.31	54.633 ^b	b
$q^{K}_{22} \cdot 10^{3}$	-10.158	4.292 ^b	b
$A\zeta \cdot 10^{-3}$	159.92059		141.90828
$\eta_J \cdot 10^3$	466.883		336.179
$\eta_{\scriptscriptstyle K}$	12.0099		11.3122
$\eta_{\scriptscriptstyle JK} \cdot 10^6$	-32.74		-51.9
$\eta_{\scriptscriptstyle KK} \cdot 10^3$	1.03		5.11
$F_{9/22} \cdot 10^{-3}$	8.5028 ^c		c
$F^{J}_{9/22} \cdot 10^{3}$	25.9 ^c		c

 $^{\mathrm{a}}\Delta X:=X_{v}$ – $X_{v=0}$.

^bBetween $v_{10} = 2$, l = 0 and $v_{10} = 2$, $l = \pm 2$. ^cBetween $v_9 = 1$ and $v_{10} = 2$, $l = \pm 2$.

	$v_{10} = 1$
$E_{\rm v}\cdot 10^{-6}$	9.921 549 62 (78)
$\Delta(A-B)$	-85.8403 (108)
ΔB	23.939204 (93)
$\Delta D_J \cdot 10^6$	77.173 (42)
$\Delta D_{JK} \cdot 10^6$	609.25 (109)
$\Delta D_K \cdot 10^3$	8.585 (117)
$\Delta H_J \cdot 10^{12}$	216.8 (79)
$\Delta H_{JK} \cdot 10^9$	-2.002 (186)
$\Delta H_{\rm KJ} \cdot 10^9$	75.9 (80)
q_{22}	-16.786365 (111)
$q^{J}_{22} \cdot 10^{6}$	53.6194 (295)
$q^{K}_{22} \cdot 10^{3}$	4.6341 (35)
$A\zeta \cdot 10^{-3}$	141.920464 (31)
$\eta_J \cdot 10^3$	334.7653 (82)
$\eta_{\scriptscriptstyle K}$	10.927 31 (185)
$\eta_{\scriptscriptstyle JK} \cdot 10^{\scriptscriptstyle 6}$	-33.490 (151)
$\eta_{\scriptscriptstyle KK} \cdot 10^3$	_
$C_{9/10}$	20.127 (233) ^b
$F_{10/22} \cdot 10^{-3}$	52.274 (76) ^c
$F^{J}_{10/22} \cdot 10^{3}$	-300.7 (67) ^c

Spectroscopic constants^a (MHz) of propyne, $H_3C-C=CH$, in its $v_{10} = 1$ vibrationally excited state, taking into account the interactions with $v_9 = 1$ and $v_{10} = 2$.

^a $\Delta X := X_v - X_{v=0}$. ^bBetween $v_{10} = 1$ and $v_9 = 1$. ^cBetween $v_{10} = 1$ and $v_{10} = 2$, $l = \pm 2$.

Spectroscopic constants^a (MHz) of propyne, H₃C-C=CH, in its v = 0 and $v_{10} = 1$ states, taking into account the interactions with $v_9 = 1$ and $v_{10} = 2$.

v = 0		$v_{10} = 1$	
Parameter	Value	Value	Parameter
		9.921 549 57 (91)	$E_{ m v}\cdot 10^{-6}$
A-B	150 594.45 (46)	-85.840 (74)	$\Delta(A-B)$
В	8 5 4 5.876 8 6 1 1 (2 1 4)	23.939 182 (103)	ΔB
$D_J \cdot 10^3$	2.939 241 4 (240)	77.171 (50)	$\Delta D_J \cdot 10^6$
$D_{JK} \cdot 10^3$	163.414 05 (172)	609.54 (138)	$\Delta D_{_{JK}}\cdot 10^{6}$
D_{K}	2.907 2 (122)	8.576 (117)	$\Delta D_{K} \cdot 10^{3}$
$H_J \cdot 10^{12}$	-63.8 (58)	216.9 (98)	$\Delta H_J \cdot 10^{12}$
$H_{JK} \cdot 10^9$	914.12 (67)	-2.048 (242)	$\Delta H_{_{JK}} \cdot 10^9$
$H_{\rm KJ}\cdot 10^6$	5.2958 (101)	75.6 (83)	$\Delta H_{\scriptscriptstyle KJ} \cdot 10^9$
$L_{JJK} \cdot 10^{12}$	-6.288 (112)		
$L_{JK} \cdot 10^{12}$	-42.70 (157)		
$L_{\rm KKJ}\cdot 10^{12}$	-395.6 (188)		
		-16.786365 (111)	q_{22}
		53.6198 (295)	$q'_{_{22}} \cdot 10^{_{6}}$
		4.6338 (36)	$q^{\kappa}_{22} \cdot 10^3$
		141.92046 (44)	$A\zeta \cdot 10^{-3}$
		334.7657 (84)	$\eta_J \cdot 10^3$
		10.927 (49)	$\eta_{\scriptscriptstyle K}$
		-33.447 (155)	$\eta_{\scriptscriptstyle JK} \cdot 10^6$
		_	$\eta_{\scriptscriptstyle KK} \cdot 10^3$
		20.081 (237) ^b	$C_{9/10}$
		52.269 (76) ^c	$F_{10/22} \cdot 10^{-3}$
		-302.7 (70)°	$F^{J}_{10/22} \cdot 10^{3}$

^a $\Delta X := X_v - X_{v=0}$. ^bBetween $v_{10} = 1$ and $v_9 = 1$. ^cBetween $v_{10} = 1$ and $v_{10} = 2$, $l = \pm 2$.

Table 5 Spectroscopic constants (cm⁻¹) of propyne, $H_3C-C=CH$, in its ground vibrational state.

Parameter	Value
Α	5.308 3500 (155)
В	0.285 059 768 32 (72)
$D_J \cdot 10^9$	98.04253 (84)
$D_{JK} \cdot 10^6$	5.450921 (70)
$D_K \cdot 10^6$	96.97 (41)
$H_J \cdot 10^{15}$	-2.107 (203)
$H_{JK} \cdot 10^{12}$	30.4905 (296)
$H_{\rm KJ}\cdot 10^{12}$	176.86 (38)
$L_{JJK} \cdot 10^{18}$	-210.6 (49)
$L_{JK} \cdot 10^{15}$	-1.394 (63)
$L_{\rm KKJ} \cdot 10^{15}$	-13.99 (68)

Table 6

Spectroscopic constants^a (cm⁻¹) of propyne, H₃C-C=CH, in its $v_9 = 1$ and $v_{10} = 2$ vibrationally excited states used for the global $v_{10} = 1$ fit.

	$v_9 = 1$	$v_{10} = 2, l = 0$	$v_{10} = 2, \ l = \pm 2$
$E_{ m v}$	638.570 562	650.862086	671.338 686
Α	5.307 015 8	5.305 249 2	5.303 053 5
В	0.285 232 61	0.286 655 09	0.286 640 31
$D_J \cdot 10^9$	98.4558	103.4396	103.1487
$D_{JK} \cdot 10^6$	5.45397	5.473 82	5.48043
$D_K \cdot 10^6$	97.75	99.84	99.63
$q_{22}\cdot 10^6$	-300.84	-559.87 ^b	b
$q^{J}_{22} \cdot 10^{9}$	0.6442	1.822 4 ^b	b
$q^{K}_{22} \cdot 10^{9}$	-338.8	143.2 ^b	b
Aζ	5.306740		4.733 551
$\eta_{\scriptscriptstyle J}\cdot 10^{\scriptscriptstyle 6}$	15.5735		11.2137
$\eta_{\scriptscriptstyle K} \cdot 10^6$	400.61		377.33
$\eta_{_{JK}} \cdot 10^9$	-1.092		-1.731
$\eta_{\scriptscriptstyle KK} \cdot 10^9$	34.3		171.
$F_{9/22} \cdot 10^3$	283.62 ^c		с
$F^{J}_{9/22} \cdot 10^{9}$	964.°		с

^aSextic and octic centrifugal distortion constants fixed to ground state values.

^bBetween $v_{10} = 2$, l = 0 and $v_{10} = 2$, $l = \pm 2$.

^cBetween $v_9 = 1$ and $v_{10} = 2$, $l = \pm 2$.

	$v_{10} = 1$
$E_{ m v}$	330.947 272 (26)
A	5.306 285 2 (4)
В	0.285 858 294 2 (31)
$D_J \cdot 10^9$	100.61673 (141)
$D_{JK} \cdot 10^6$	5.471 243 (35)
$D_K \cdot 10^6$	96.688 (4)
$H_J \cdot 10^{15}$	5.125 (264)
$H_{JK} \cdot 10^{12}$	30.4237 (62)
$H_{\rm KJ}\cdot 10^{12}$	179.39 (27)
$q_{22}\cdot 10^6$	-559.9329 (37)
$q'_{22} \cdot 10^9$	1.78855 (98)
$q^{K}_{22} \cdot 10^{9}$	154.576 (116)
Aζ	4.733 957 12 (102)
$\eta_{\scriptscriptstyle J}\cdot 10^{\scriptscriptstyle 6}$	11.166568 (274)
$\eta_{\scriptscriptstyle K}\cdot10^{\scriptscriptstyle 6}$	364.496 (62)
$\eta_{_{JK}} \cdot 10^9$	-1.1171 (50)
$\eta_{\scriptscriptstyle KK} \cdot 10^9$	_
$C_{_{9/10}}\cdot 10^{_{6}}$	671.4 (78) ^b
F _{10/22}	1.743 69 (255) ^c
$F^{J}_{10/22} \cdot 10^{6}$	-10.029 (225) ^c

Spectroscopic constants^a (cm⁻¹) of propyne, H₃C-C=CH, in its $v_{10} = 1$ vibrationally excited state, taking into account the interactions with $v_9 = 1$ and $v_{10} = 2$.

^aOctic centrifugal distortion constants fixed to ground state values. ^bBetween $v_{10} = 1$ and $v_9 = 1$.

^cBetween $v_{10} = 1$ and $v_{10} = 2$, $l = \pm 2$.